



Applicability Assessment of Oil Palm Trunk Biochar for Use as Soil Amendment: Morphology, Structure, and Chemical Properties

Ebsan Marihot Sianipar^{1,2,*}, Sumihar Hutapea³, Ernitha Panjaitan², Siti Mardiana³

¹*Program in Agriculture Science, Universitas Medan Area, Medan 20112, Indonesia*

²*Department of Agrotechnology, Faculty of Agriculture, Universitas Methodist Indonesia, Medan 20132, Indonesia*

³*Department of Agrotechnology, Faculty of Agriculture, Universitas Medan Area, Medan 20112, Indonesia*

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ABSTRACT

This study was conducted to investigate the quality of biochars derived from oil palm trunk (OPT) based on relative trunk height (bottom, middle, top). The OPT biochars were produced by drum retort kiln with a temperature range of 300-400°C (slow pyrolysis conditions). Morphology, structure, chemical properties and heavy metal content of the OPT biochars were measured in order to assess their suitability for use as a soil amendment. The synthesized carbon was analyzed by X-ray diffraction analysis (XRD). The morphology of the OPT biochars was analyzed by scanning electronic microscopy (SEM) coupled with an energy dispersive X-ray spectrometer (EDX) to identify mineral species on the OPT biochar surface. The analytical methods applied for biochar characterization were proximate analysis and elemental analysis. Characterization of surface functional groups of the OPT biochar was carried out using Fourier transform infrared spectroscopy (FTIR). According to the analysis results, the biochar derived from bottom OPT had the highest intensity peak measured by XRD; it contained 62.05% fixed carbon, 69.21% carbon, and it had the lowest oxygen content at 26.28%. The highest number of pores was found in biochar derived from top OPT. Overall, the biochars had rich macronutrients, numerous functional groups, and low heavy metal content. This study showed the applicability of oil palm trunk biochar for use as a soil amendment for agricultural applications.

Keywords: Carbon; Characterization; Drum retort kiln; Heavy metal; Pyrolysis

1. Introduction

Currently, biomass is one of the main renewable and sustainable bioenergy sources that is already widely used around the world [1]. Biomass can also be used to produce different chemicals and materials [2, 3] that can be converted into three distinct forms of bioenergy fuels, those being liquid (bio-oil), gaseous (syngas) and solid (biochar) forms [4-7]. However, the potentiality of biomass as feedstock in each country or region depends on a variety of factors, such as location, climate, weather, available local plantations, agricultural activity, and industrial processing. For example, Indonesia has many agro-industries, which has led to the production and use of various types of biomass.

Indonesia became the global leader in palm oil export, with the increase in the plantation area from 0.3 million hectares in 1980 to 14.62 million hectares in 2021 [8]. The expansion of oil palm plantations has been identified as a significant source of anthropogenic greenhouse gas (GHG) emissions [9]. Meanwhile, the harvesting, processing, and replanting of oil palm produces many forms of oil palm biomass (OPB), including empty fruit bunches (EFB), oil palm fronds (OPF), mesocarp fiber (MF), oil palm kernel shells (PKS), palm oil mild effluent (POME), and oil palm trunks (OPT) [10]. It has been estimated that by 2030, Indonesia will produce 54 million tons of EFB, 115 million tons of OPF, 31 million tons of MF, 15 million tons of PKS, 130 million tons of POME, and 59.7 million tons of OPT [11]. Utilization of oil palm biomass in all of these forms is a promising way to mitigate the negative impacts of oil palm plantations and to develop renewable and sustainable bioenergy sources.

Across Indonesia, there are 14.3 million hectares of oil palm plantations with a 4%/year rejuvenation rate [13]. In the context of commercial oil palm production, the life cycle of the oil palm plant requires that it be replaced every 20 to 25 years to

maintain the desired level of oil production [12]. During the process of replanting oil palm trees, large quantities of OPT and OPF waste are produced. It has been estimated that this replanting generates, on a dry weight basis, 14.4 ton /ha of OPF and 66 ton/ha of OPT [14]. Further, there is about 40.0 million tons/year of trunk waste production. Oil palm trunk, a biomass source that possesses the desirable property of fast growth, is an alternative to native or reforested wood. OPT remains an underutilized byproduct of palm oil, which is often shredded in the field at the end of the productive lifetime of a plantation tree. Therefore, OPT is one of the most promising sources of feedstock biomass for biochar synthesis. Biochar is a carbon-rich byproduct of thermal degradation of organic materials under an oxygen-depleted environment (i.e., pyrolysis) and was recently recognized as an emerging technology and is distinguished from charcoal by its ability to be used as a soil amendment [15-18]. However, characteristics of biochar vary significantly, depending on feedstocks, production methods, and temperatures [19, 20].

OPT, is mainly composed of hemicelluloses, cellulose, and lignin and has a high potential for use in biochar applications in Indonesia [21]. In terms of sustainably, it is assumed that the conversion of OPT biomass into other forms of biochar is much more effective and useful than the direct burning of biomass. Direct burning produces a large amount of pollutants that are hazardous to both human health and the environment. At present, the use of OPT to produce biochar using pyrolysis-based processes is still a challenge as there have only been a few studies that have investigated this process as it relates to trunk height [22-29], necessitating further studies to obtain in-depth insights into the qualities of biochars derived from bottom, middle, and top sections of OPT. These three sections of OPT biomass were pyrolyzed using a drum retort kiln. It is crucial to verify the

morphology, structure, chemical properties, and heavy metal content of OPT biochars based on trunk section. The different trunk sections of OPT (bottom, middle, top) were assessed for variations in composition as it relates to the suitability for use as a soil amendment. We performed a full characterization of the material based on the International Biochar Initiatif (IBI) Biochar Standards [30] and the European Biochar Community (EBC) Standards [31]. The purpose of this research was therefore to investigate the quality of biochars derived from bottom, middle, and top sections of OPT for the suitability of the biochar as a soil amendment.

2. Materials and Methods

2.1 Material

OPT was collected from oil palm plantation PTPN IV at Tanah Jawa, Simalungun District, North Sumatera Province, Indonesia. OPT was harvested at the oil palm plantation, selecting mature stem samples at 23 years of age, taking cylinder logs approximately 15 min length and 59 cm in diameter. OPT is fibrous, bulky, and has a high moisture content of about 76%. OPT sampling was conducted by sectioning the trunk into three parts: bottom, middle and top. The boundaries of the three trunk sections were 0-30%, 30-60%, and 60-100% of the total height, corresponding to bottom, middle, and top, respectively. First, the OPT was chipped by heavy-duty chipper into pieces about 10-15 cm thick (Fig. 1). The OPT was then sun-dried for 6 hours a day for ten days to reduce the moisture content to around 15% by weight (Fig. 2). The higher the water content of the feed-stock, the more combustion energy is needed to evaporate the water and to heat the feedstock to pyrolysis temperatures (300-400°C).



Fig. 1. OPT chipped by heavy equipment.



Fig. 2. Sun dried of OPT chipp.

2.2 Pyrolysis

Pyrolysis was carried out using a drum retort kiln with a 200 liters drum reactor system. The center of the drum had a hole at top (cap) and bottom (floor) where a 6 in diameter pipe was placed. The pipe had holes about 10 cm from its bottom to allow smoke to escape. Dried OPT biomass was placed in the drum retort. The biomass was heated directly by burning wood in the kiln. The drum was sealed by heat isolation fabric (Fig. 3) which allowed the drum to form the necessary gas-tight seal required for carbonization to take place.



Fig. 3. The drum retort kiln.

The kiln was fired and the biomass was heated for 7 hours, reaching temperatures of 300°C to 400°C. The temperature reading was shown on digital infrared thermometer (BNQ, BN 1000) (Fig. 4). After cooling for 12 hours, the drum was opened and the biochar was removed.



Fig 4. The temperature reading.



Fig. 5. OPT biochar.

2.3 Characterization of biochar

The biochar (Fig. 5) were ground to particle size < 20 mess and three representative samples (bottom, middle, and top OPT biochar) were taken for analysis. The crystallography of OPT biochar was characterized using X-ray diffraction analysis (XRD) SHIMADZU Lab X 6100 Japan. X-ray pattern was equipped with Cu K α radiation ($\lambda = 1.5406 \text{ \AA}$) with a targeted voltage of 45 kV and a current of 40 m \AA . The scans were collected from 5-70° and used a step size of 0.02° 2 θ and a scan speed of 1.0° 2 θ /min. The morphology and the mineral species of the biochar were identified using scanning electron microscopy (SEM) (ZEISS EVO MA 10 Germany) and energy dispersive X-ray spectroscopy analysis (EDX) (BRUKER 129 eV Germany), with 15 kV and 180 Pa for acceleration voltage

and beam current, respectively, in a vacum of 25 Pa with an Au coating. The analytical methods applied for biochar characterization were proximate analysis and elemental analysis. The proximate analysis for moisture (Carbolite minimum free space oven), ash (Carbolite Horizontal Muffle Furnace), and volatile contents (Ishizuka Denki Muffle Furnace) were determined according to the ASTM D1762-84 standard method. Fixed carbon was determined according to ASTM D3172-17. The elemental analysis (CHN) was performed in duplicate using an Elemental Analyser (Leco CHN 628 USA), and element S using Leco S144-DR. The organic functional chemical groups of biochar were identified using the fourier transform infrared spectroscopy (FTIR), Nicolet IS 10 ATR USA (38/IKA/MT). The chemical functional groups in biochar are vital to understanding the chemical characteristics of the biochar produced. The spectra were recorded with a 4 cm $^{-1}$ resolution between wave numbers of 4000 $^{-1}$ and 500 cm $^{-1}$. The graph is drawn using Origin 8 pro, Excel. Heavy metal content analysis was performed using HNO₃ digestion (CEM Mars 6) and followed by determination with Atomic Absorption Spectrophotometry (Hitachi ZA 3000, Japan).

3. Results and Discussion

3.1 XRD analysis

The result of the XRD measurement of the synthesized carbon of the OPT biochars is shown in wide-angle range (5° to 70°) (Fig. 6). In the case of the carbon, it demonstrated a broad peak at 2 θ value of 21-22°, which corresponds to the amorphous plane of (002) carbonaceous materials [32]. The spectra of the C-OPT biochars presented characteristic peaks at 21.92°, 44.12°, and 64.62° for the biochar derived from bottom OPT. As for the biochar derived from middle OPT, the characteristic peaks were at 21.72°, 44.3° and 64.22°. While the characteristic peaks were at 22.5°, 44.08°, and 64.22° for the biochar

derived from top OPT corresponding to the (002), (200) and (220) planes of carbon, respectively. The highest intensity of the peaks was shown by biochar derived from the bottom OPT, followed by middle OPT, indicating that the bottom OPT biochar has the highest degree of graphitization [33], and the broad peaks of OPT biochar also suggest that the synthesized carbon from OPT is amorphous [34]. The thermal pyrolysis enabled the trans-formation of biomass to biochar by condensation of smaller organic

molecules into conjugated aromatic rings, meanwhile producing massive defective edges along the biochar boundaries terminated with hydrogen atoms and oxygen functionalities [35, 36]. The quality of biochar is determined by its fixed carbon content and biochar composition is crucial to define its application [37]. One identified option for sustainable soil management practice is to increase soil organic carbon levels, especially with recalcitrant forms of carbon (e.g., biochar application) [38].

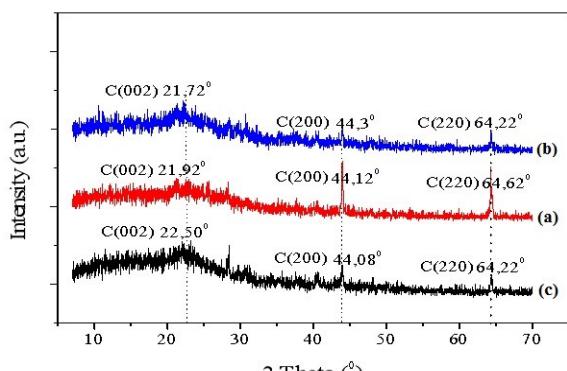


Fig. 6. XRD pattern of the OPT biochars: (a) bottom; (b) middle; (c) top.

3.2 Morphological analysis

The SEM-EDX analysis of OPT biochar is presented in Fig. 7. The morphology of top-OPT biochar had relatively more pores on its surface. In the case of bottom-OPT biochar, the pore size was smaller and the number of pores was fewer (Fig 7a). This is because the bottom-OPT biochar contained the highest level of lignin, giving it a high density structure that was hard and solid. Consequently, the pyrolysis of the bottom OPT did not completely devolatilize most of lignin, and the pores were not fully developed. The formation of pores and an increase in surface area were mostly due to the removal of moisture and certain volatile matter due to the elimination of hemicelluloses and cellulose. Further increase in pore size, pore

volume, and surface area was observed as mostly the major constituents of lignocelluloses, i.e., lignin, cellulose, and hemicelluloses, were broken down as depicted in previous studies [39]. As a result, the highest number of pores was observed in the top-OPT biochar (Fig. 7c). Porous channels and a higher amount of mesopores were observed on the surface of the OPT biochar produced in this study, similar to previous results [40-42]. The synthesis and release of volatile molecules from OPT throughout the carbonization process is indicated by porous channels. The remaining non-volatile components are subsequently converted into biochar, which has pores of various sizes and shapes visible on the surface.

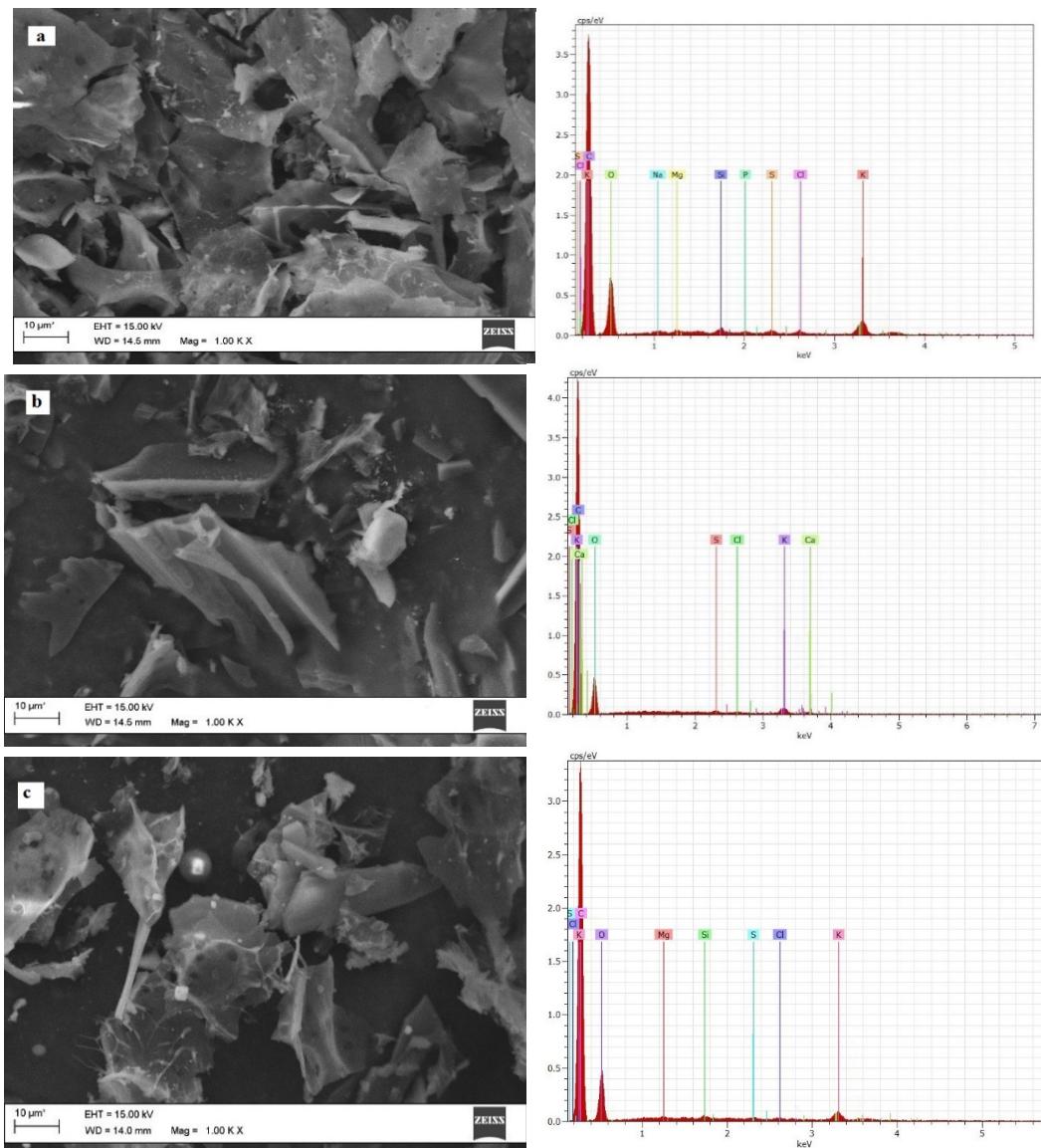


Fig. 7. SEM photographs at 1,000X and EDX analysis of the surface of the biochar: a. bottom; b. middle; c. top.

The large pores are caused by the progressive degradation of the lignocellulosic components including cellulose, hemicelluloses, and lignin. At higher temperatures, these pores also facilitate the release of volatile materials resulting in larger pore size [40]. Furthermore, the highly porous structure of biochar may be beneficial for biofilter applications and soil mixing [44]. This highly porous structure increases water

holding capacity by increasing the total porous space of the soil, in addition to its benefits as a soil amendment [45].

On the surface of the OPT biochars, EDX examination demonstrated the coexistence of elements C, O, K, Ca, Mg, Si, Cl, S, and P. This is because the feedstock (OPT) contains bioavailable elements that have an impact on the soil environment. Carbon and oxygen are prominent on all surfaces, with carbon

constituting almost the entire surface composition. Minor peaks were also observed for Mg, K, Si, Cl, S, and P on all surfaces. Low pyrolysis biochars showed higher concentrations of Ca, Mg, K, Fe, Mn, and Zn than fast pyrolysis biochars did [46]. This is in agreement with another study, which reported that K, Ca and Mg appeared to be retained in low pyrolysis temperatures of 400°C [47]. The biochar ash mineral content includes inorganic constituents such as oxides, carbonates, silicates, sulfates, chlorides, and phosphates of some metals [48]. The macronutrients P, K, S, Ca, and Mg are among those found in OPT biochars. The type of feedstock and the processing conditions have a significant impact on the nutritional content of biochar. One of the main oxides found in OPT is K₂O, along with CaO, SiO, P₂O₅, and MgO [49]. Nutrient availability is related to the nature of the chemical compounds in which the key elements occur [50].

Table 1. Proximate and ultimate analyses of OPT biochar produced by drum retort kiln.

Property	Bottom	Middle	Top	Average \pm St.Dev
Proximate Analysis (% dry basis)				
Moisture	4.54	5.22	6.07	5.28 \pm 0.77
Fixed Carbon	62.05	57.93	54.34	58.11 \pm 3.86
Volatile Matter	30.12	31.23	33.32	31.56 \pm 1.62
Ash	7.83	10.84	12.34	10.34 \pm 2.30
Elemental Analysis (% dry basis)				
C	69.21	67.31	65.49	67.34 \pm 1.86
H	4.42	4.79	4.43	4.55 \pm 0.21
N	0.06	0.05	0.05	0.05 \pm 0.01
S	0.03	0.04	0.04	0.04 \pm 0.01
O*	26.28	27.49	29.99	27.92 \pm 1.89
Bulk atomic ratios				
H/C	0.76	0.84	0.81	0.80 \pm 0.04
O/C	0.38	0.41	0.46	0.42 \pm 0.04
(N + O)/C	0.38	0.41	0.46	0.42 \pm 0.04

The remaining volatile matter in biochar includes oxygenated compounds such as acetic acid, furans, phenols, and 2-Propanone,1-hydroxy- [40]. In the present study, it was observed that the moisture content of biochar samples was not zero after pyrolyzing, which is similar to results from previous studies [40, 41].

The quality of biochar is determined by its fixed carbon content and the

3.3 Proximate and elemental analysis

The proximate and elemental analysis results of OPT biochar are summarized in Table 1. The relative standard deviation for the proximate analysis was < 4%, while it was < 2% for the ultimate analysis. The proximate analysis showed that the moisture content, fixed carbon content, volatile matter, and ash content of the OPT biochar derived bottom to top trunk were in the ranges of 4.54-6.07, 62.05-54.34, 30.12-33.32, and 7.83-12.34 (wt.%), respectively. The data suggest that there is a small difference between the biochar derived from bottom, middle and top OPT. The highest fixed carbon content was found in bottom-OPT biochar (62.05%), followed by middle-OPT biochar (57.93%), and the lowest content in top-OPT biochar (54.34%). However, the volatile matter of the bottom-OPT biochar was the lowest of the three, at 30.12 wt.%.

composition of biochar is crucial in determining its potential applications [20]. The proportion of fixed carbon was related to the presence of stable aromatic carbon in the biochar [51]. Taking into account the recalcitrant nature of carbon contained in the biochar (high resistance to mineralization due to the microbial action), the above result suggests that the pyrolysis of OPT is an effective way to increases the

turnover time of the carbon contained in the biomass. The high fixed carbon content of biochar is favorable for use as a soil amendment [40, 52]. The high fixed carbon content of biochar also helps improve soil carbon sequestration that results in an increase in soil organic carbon content [53]. Biochar soil management systems could deliver tradable carbon emissions reduction as the carbon sequestered could be the basis for sustainable oil palm cultivation, even with the impacts of climate change in Indonesia. This could combat desertification, sequester atmospheric CO₂ in the long term, and help to maintain biodiversity in tropical rainforests [54]. However, the volatile matter of the biochar remained between 30.12-33.32 wt.%, indicating incomplete pyrolysis of the biomass under the investigated pyrolysis temperatures and time. This is because the biomass samples contained some lignin, which only decomposes at high temperatures. The lower volatile matter led to a higher fixed carbon content in the biochar. The ash content of OPT was 7.83-12.34 wt.%. The ash in biochar is non-volatile and non-combustible. The increase in ash content of biochar resulted from the destructive volatilization of lignocellulose components at higher temperatures [55]. Lignin, with its greater chemical stability, can be only partially degraded during pyrolysis. Correspondingly, the original skeleton of the particles is mostly preserved [56] and the particles retain the cellular appearance of the raw materials [57].

The elemental composition of the OPT biochar in regards to C, H, N, S, and O content derived from bottom to top trunk were in the ranges of 69.21-65.49; 4.42-4.79; 0.05-0.06, 0.03-0.04 and 26.28-29.99 wt.%, respectively. The results show that carbon and oxygen are the most abundant elements present in the OPT biochar, whereas hydrogen, nitrogen, and sulphur were detected at low concentrations. It should be noted that the feedstock (oil

palm) has been found to store an estimated 37.8 to 42.1 t C ha⁻¹ in its above-ground biomass [58]. Biochar is graded based on its carbon content, split into three classes. Class I has a carbon content of 60% and higher, class II has a carbon content of 30 to 60%, and class III has a carbon content of 10 to 30% [30]. According to IBI biochar standards, the organic carbon content of the OPT biochars in this study were all > 60 %, placing them in class I. The results of ultimate analysis were consistent with the proximate analysis results as indicated by the relation between fixed carbon content and volatile matter, and carbon and hydrogen content. These results clearly showed that the pyrolysis of biomass produced biochar with a high carbon content and a low oxygen content, which is similar to the results from previous studies [40]. This biochar product is suitable to be used as a soil amendment to improve soil fertility [59].

As one of the roles of biochar is to store carbon, the C yield and H/C ratio is more important than the biochar yield in regards to carbon sequestration [60]. Generally, the H/C molar ratio can be used as an indicator of the degree of aromaticity and carbonization. Combustion analysis H/C ratios of 0.76 (bottom), 0.84 (middle) and 0.81 (top) OPT biochar indicate similar aromaticities. An H/C ratio > 0.6 indicates the possibility of not being fully carbonised [52]. This means that a low H/C ratio indicates a biochar that highly carbonised, exhibits a highly aromatic structure, and has a generally high stability for storing carbon [61]. Furthermore, the H/C and O/C ratios are essential parameters for stability and degree of oxidation in the biochar structure. Biochar produced at low temperatures will have O/C and H/C ratios that are larger than those of biochar produced at high temperatures [62].

Biochars with low H/C and O/C ratios are graphite-like materials or charcoal which are highly stable compared

to their original biomass feedstock having higher H/C and O/C ratios [31]. The polarity index ((O + N)/C), an indication of the surface hydrophilicity and polar group content, was similar for bottom-OPT biochar (0.38), middle-OPT biochar (0.41) and top-OPT biochar (0.46) indicating similar hydrophilicities. Essentially, every carbon-rich solid often consists of two parts: hydrophobic core (a highly aromatic

nucleus) and hydrophilic shell (outer layer; a high concentration of reactive oxygen functional groups like hydroxyl/phenolic, carbonyl, or carboxylic groups) [63].

3.4 FTIR analysis

FTIR analysis was conducted to determine the functional groups in OPT biochars (Fig. 8).

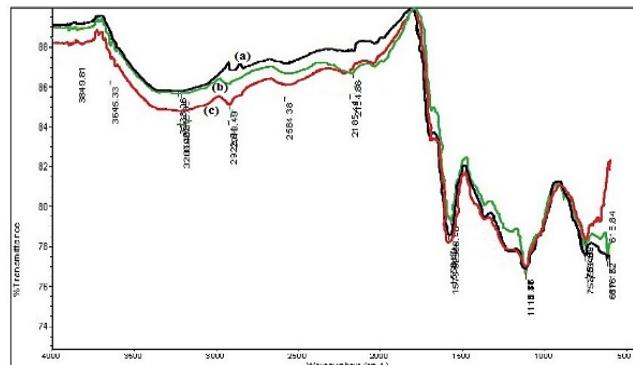


Fig. 8. FTIR spectra of OPT biochar: (a) bottom, (b) middle, (c) top.

Fig. 8 showed that the classification of chemical compounds present in the biochar can be derived from the functional groups that were detected as vibrational modes (or transmittance peaks) observed at

different ranges of wavenumber (Table 2). The OPT biochars had several bends which indicates that they had a high level of functional groups present at the surface.

Table 2. The FTIR results for OPT biochar.

Frequency range (cm ⁻¹)	Functional groups	Bottom	Middle	Top
3200-3500	O-H stretching	3233	3215	3207
2800-2950	C-H stretching	2892	2918	2919
2500-2700	Carboxylic O-H	2601	2575	2577
2200-2500	C=C stretching	2089	2164	2229
1600-1700	Aldehid C=O	1697	1695	1694
1583-1464	C=C stretching	1581	1568	1583
1371-1250	C-O stretching	1363	1371	1371
1100-1212	C-H bending	1212	1112	1114
750-600	C-H in plane bending	752	751	754

This is because the biochar still contained structures of carbon, hydrogen, and oxygen, which mostly remained in the form of lignin. The results of FTIR analysis are consistent with the proximate and ultimate analysis results discussed previously. The spectra of the three OPT biochars did not differ because their

elemental compositions were relatively similar as indicated by the content of carbon, hydrogen, and oxygen. The peaks can be explained as follows. The first peak appeared at 3200-3600 cm⁻¹ and was attributed to the stretching of an O-H group [63, 64]. It is also attributed to the acceleration in the dehydration reaction of

biomass. The small peaks at 3208 cm⁻¹ were associated with the C-H stretching vibration of aliphatic and aromatics structures. The carboxylic O-H occurred in the range 2500-2700 cm⁻¹ [65]. The observed peak at 1694-1697 cm⁻¹ is attributed the presence of a carbonyl group in carbohydrate [66]. The aromatic C = C ring stretching vibration occurred at 1568-1583 cm⁻¹ [43, 65]. At 1371 cm⁻¹, the peak is assigned mainly to stretching vibrations of aliphatic C-H and CH₂ bending in biochar. The band in the

range of 1115 cm⁻¹ represents the stretching of aromatic C-O and phenolic OH. The weak vibrations of the C-H bond in aromatic and heteroaromatic compounds are visible as a band between 604-754 cm⁻¹ [66, 67]. Among these, the carboxylic and phenolic groups were the primary acidic functional groups. These groups provide an opportunity for application in the immobilization of heavy metals in the soil [68].

Table 3. Heavy metal content of OPT biochar.

Element	Bottom	Middle	Top	Average ± St.Dev	(EBC Standards)[31]	(IBI Biochar Standards) [30]
				mg kg ⁻¹		
As	0.03	0.10	0.10	0.08 ± 0.04	< 13	12-100
Cd	nd	nd	Nd		< 1.50	1.4-39
Cr	6.10	4.90	8.60	6.53 ± 1.89	< 90	64-1200
Co	1.20	1.70	4.20	2.37 ± 1.61	-	40-150
Cu	84.00	87.00	26.00	65.67 ± 34.39	< 100	63-1500
Pb	1.10	1.10	6.00	2.73 ± 2.83	< 120	70-500
Hg	0.01	0.01	0.01	0.01 ± 0.00	< 1.00	1-17
Mo	nd	5.30	15.00	10.15 ± 6.86	-	5-20
Ni	0.90	1.20	2.30	1.47 ± 0.72	< 50	47-600
Se	0.60	0.40	0.40	0.47 ± 0.12		2-36
Zn	177.00	477.00	410.00	354.67 ± 157.47	< 400	200-7000

3.5 Heavy metal content

Heavy metal content and maximum allowed thresholds of IBI Biochar Standards are listed in Table 3. These thresholds were used as a reference to understand the levels of heavy metals in OPT biochars for soil amendment. Considering the heavy metals contained in the biochars, Zn had the highest concentration at 177-477 mg kg⁻¹, followed by Cu with a concentration of 26-84 mg kg⁻¹. Other heavy metal concentrations were below 15 mg kg⁻¹. According to the standards of IBI biochar thresholds [30] and the European biochar certificate [31], these results indicate that the heavy metal contents are acceptable and the biochar product is clean. Heavy metal contamination is a worldwide problem and anthropogenic activities are to blame for the increased concentrations of heavy metals in soils [70]. High concentrations of heavy metals adversely affect soil quality and biological functions due to their toxicity and persistence

after entering the soil [69]. The content of heavy metals in biochar is greatly influenced by the source biomass [71]. Overall, heavy metal content in OPT biochars do not pose threats to the environment when biochar is used as a soil amendment.

4. Conclusion

The biochar derived from the bottom section OPT provided the highest intensity of peaks and amount of fixed carbon (62.05%). Overall, each of the OPT biochars had an average carbon content > 60% making them all class I biochars, indicating that they are conducive to carbon sequestration when used as a soil amendment. The biochars all had an H/C ratio > 0.6 which indicates the possibly of not being fully carbonized. The biochars also had a porous structure, and contained macronutrients such as P, K, S, Ca, and Mg for plant growth, and oxygen containing functional groups present on the surface for the immobilization of heavy metals in the

soil. Most concentrations of heavy metals in the OPT biochars were still up to the European biochar certificate and the IBI biochar threshold standards. Based on these results, it can be concluded that OPT biochar possesses suitable properties for its use as a soil amendment for agricultural applications. Still, the research on OPT biochar as a soil amendment is scarce and more research should be conducted to further validate the benefits of it as a soil amendment on oil palm plantations.

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