



Preparation of Lead-Free X-Ray Shielding Materials Based on Natural Rubber/Barium Sulfate Composites

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Abstract: This work prepared the lead-free X-ray shielding materials based on natural rubber (NR) mixed with BaSO₄ particles at different BaSO₄ contents (0–100 phr) and different material thicknesses (0.5–2.5 cm). The X-ray shielding properties of NR/BaSO₄ composites were tested using a 100-kV X-ray generator. The results suggested that the increases in the BaSO₄ contents improved the x-ray shielding properties of the NR/BaSO₄ composites, such as linear attenuation coefficient (μ_l), mass attenuation coefficient (μ_m), and half-value layer (HVL). The highest X-ray attenuation was 94 % in 100 phr BaSO₄. In terms of mechanical properties, it was found that the density and hardness (Shore A) increased with increasing BaSO₄ contents. In contrast, the increases in BaSO₄ resulted in decreased tensile strength, elongation at break and tear strength. In summary, the overall properties suggest that these NR/ BaSO₄ composites could be developed as an effective X-ray shielding material instead of lead.

Keywords: Natural rubber; Barium sulfate; X-ray shielding; Composite materials

1. Introduction

X-ray is a relatively high-energy electromagnetic wave that plays important roles in various applications such as agriculture, food preservation, industry, and especially in medical fields for diagnosis and therapy [1]. However, exposure to X-rays could potentially lead to fatal health hazards in radiation workers and users and cause symptoms such as skin burns and vomiting within hours. It can also result in long-term health effects such as memory loss, cancer, and cardiovascular disease [2]. Therefore, effective shielding materials were important to prevent excessive X-ray exposure. It is known that the materials composed of high-atomic-number (Z) and high-density elements are commonly used as shielding materials, such as lead (Pb) [3-5].

However, lead is considered a hazardous chemical [6]. Recently, the radiation shielding materials with the addition of fillers such as tungsten oxide (W_2O_3) [7-8], bismuth oxide (Bi_2O_3) [8-10], and barium sulfate ($BaSO_4$) [7,11-12] were studied for replacement of lead (Pb). Among these materials, $BaSO_4$ is an exciting filler since it is environmentally friendly, safe, and highly effective in X-ray attenuation [12]. Previous investigations reported that it could be improved the radiation shielding properties of concrete [13] and it is also found that a shielding film prepared from $BaSO_4$ and polymeric silicon has higher tensile strength and X-ray shielding performance [7]. In addition, a polymeric composition consisting of ethylene propylene diene monomer (EPDM), industrial oil (IO), and alkyl phenol-formaldehyde resin (PF) with the addition of $BaSO_4$ can be applied to manufactured material for protection from radiation in offices of radiation therapy and diagnostics [12].

In this work, we aimed to develop flexible and lead-free X-ray-shielding materials that are safe and environmental-friendly based on NR/ $BaSO_4$ composites. Dunlop method is used for NR/ $BaSO_4$ composites foam production. This method is widely used because it is simple, convenient, reliable, cost-effective, and energy-efficient [14-15]. The X-Ray shielding properties are tested based on the thicknesses and $BaSO_4$ contents of NR/ $BaSO_4$ composites, such as X-ray transmission ratio, X-ray attenuation, linear attenuation coefficient (μ_l), mass attenuation coefficient (μ_m), and half-value layer (HVL). Furthermore, the mechanical properties of the composites were also investigated.

2. Materials and Methods

1. Materials

NR latex with 60 wt% dry rubber was acquired from the Center of rubber technology for community, Faculty of engineering, Thaksin University, Phatthalung, Thailand. Barium Sulfate was purchased from Scitrader Co, Ltd. (Thailand). Sulfur (S), zinc diethyldithiocarbamate (ZDEC), zinc 2-mercaptobenzothiazole (ZMBT), Wingstay®-L, zinc oxide (ZnO), Diphenylguanidine (DPG), and Sodium silicofluoride (SSF) was purchased from VPK Supply Co, Ltd. (Thailand). All chemicals were used in an aqueous dispersion form, prepared in the laboratory (Table 1).

2. Preparation of NR/ $BaSO_4$ composites

NR/ $BaSO_4$ composites were fabricated using the chemical proportions given in Table 1. First, NR latex was filtered and agitated by a whisk and bowl mixer for 3 min to liberate dissolved ammonia before foam preparation. After that, $BaSO_4$, S, ZDEC, ZMBT, Wingstay® L, ZnO, DPG and SSF were slowly added to the mixture and stirred for about 1 min. Then, the NR/ $BaSO_4$ foam was poured into a mold ($25 \times 25 \times 0.5 \text{ cm}^3$) to begin the gelling process. Finally, NR/ $BaSO_4$ was vulcanized at 100°C for 1 h in an oven (ED 56, BINDER). The non-reactive reagents were removed by washing the solid foam with distilled water. After washing, the cured NR/ $BaSO_4$ was evenly dried in a hot air oven at 80°C for 24 h.

Table 1. Material formulations for Natural Rubber/Barium Sulfate Composites preparation

Ingredients	Total solid content (%)	Dry weight (phr*)	Wet weight (g)
HA latex	60	100	167
S	50	2	4
ZDEC	50	2.5	5
ZMBT	50	2	4
Wingstay® L	50	1	2
ZnO	50	5	10
DPG	40	2	5
SSF	20	0.5	2.5
$BaSO_4$ (phr)	0, 50, 75, 100		

* phr is the ratio unit of weight part of chemicals to 100 wt parts of rubber

3. Characterizations

3.1. Density and morphology studies

For all NR/BaSO₄ composites, the density was measured by comparative measurement of each sample mass in air and water using a digital scale (BSA Series, Sartorius). The experimental setup showed in Figure 1. The samples with dimensions of 20 × 20 × 5 mm³ were placed in the balance pan and recorded their dry weight. Next, the samples were placed under the submersed basket. Finally, the specific gravity was calculated by a digital scale. The specific gravity of the sample was multiplied by water density to obtain the density of the sample, as shown in Eq (1).

$$\text{specific gravity} = \frac{\text{density of the object}}{\text{density of the water}} \quad (1)$$

For the study of morphology, the samples were investigated using a scanning electron microscope (SEM; FEI / FEG Quanta 450), operating at 5 kV. Dried foams were vacuum-coated with gold before SEM investigation.

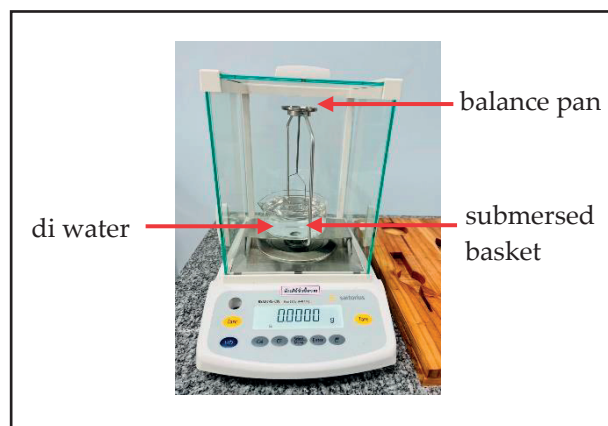


Figure 1. Setup for Density measurement.

2.2. Mechanical properties

Tensile and tear tests were conducted following ISO37-1 and ISO30-1, respectively, to characterize the mechanical properties of the NR/BaSO₄ composites. The tests were performed on a universal testing machine (Purchem Co., Ltd.) with a crosshead speed of 50 cm/min. Ultimate tensile strength, elongation at break and tear strength were determined. The hardness for all NR/BaSO₄ was tested using an automatic hardness tester (GX-02, Teclock).

2.3. X-Ray shielding properties

The X-Ray shielding properties of NR/BaSO₄ composites were performed at the Chaiya Hospital, Surat Thani Province, using the General X-ray Machine (MRAD-A32S, CANON) at 100 kV and the X-ray test equipment (X2, RAYSAFE) was used as the detector. All samples used in the experiment were 2 cm × 2 cm. The X-Ray shielding properties test was carried out on the different material thicknesses (0.5-2.5 cm). The narrow beam X-ray source was located 1 m from the surface of NR/BaSO₄ composites. The independent 0.05 s tests were performed and were recorded for each sample. The experimental setup is shown in Figure 2. From the experiment, the X-Ray shielding properties included X-ray attenuation [2], linear attenuation coefficient (μ_l), mass attenuation coefficient (μ_m), and half-value layer (HVL) [7] could be calculated using equation (2) – (5)

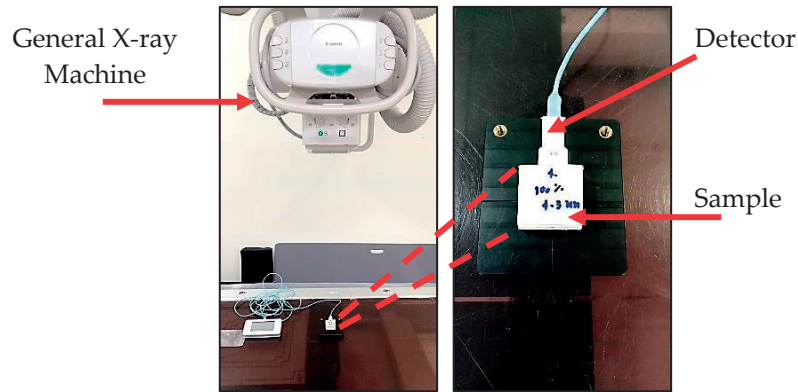


Figure 2. Setup for X-ray shielding measurement.

$$X - ray \text{ attenuation} = \frac{I_0 - I}{I_0} \times 100 \quad (2)$$

$$I = I_0 e^{-\mu_l x} \quad (3)$$

$$\mu_m = \frac{\mu_l}{\rho} \quad (4)$$

$$HVL = \frac{\ln 2}{\mu_l} \quad (5)$$

Where I , I_0 , μ_l , μ_m , ρ , HVL, and x was the intensity of transmitted X-rays, the intensity of incident X-rays, linear attenuation coefficient, mass attenuation coefficient, density, half-value layer, and thickness of the samples, respectively.

3. Results and Discussion

3.1 Density and morphology studies

The density of NR/BaSO₄ composites is shown in Figure 3. It was found that the densities of NR/BaSO₄ composites increased with BaSO₄ content mainly due to the much higher density of BaSO₄ particles. It was known that increased material density could improve the mass attenuation coefficient according to equation 4.

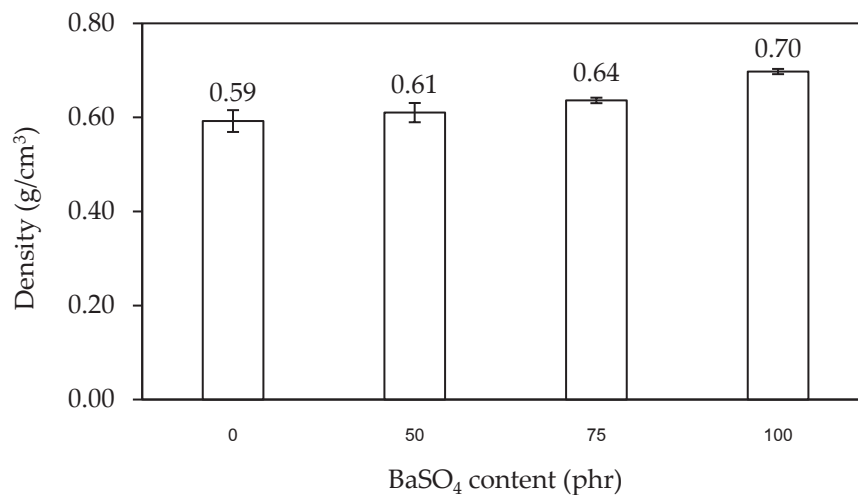


Figure 3. The densities of NR/BaSO₄ composites with varying contents of BaSO₄.

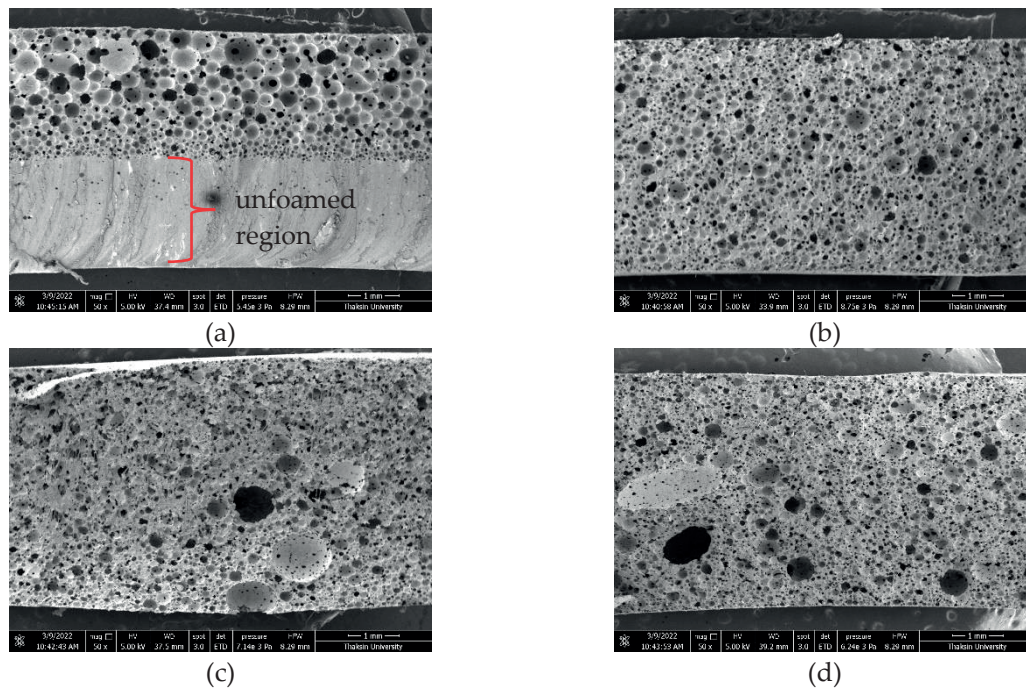


Figure 4. Cross-sectional microstructures of NR/BaSO₄ composites with varying contents of BaSO₄ (a), 0 phr, (b) 50 phr, (c) 75 phr, and (d) 100 phr.

Figure 4. represents the influence of BaSO₄ on morphologies of NR/BaSO₄ composites. From cross-sectional microstructures, All samples showed small open-cell structures inside NR/BaSO₄ composites from cross-sectional microstructures. However, NR/BaSO₄ composites had non-uniform cell structures. The composites without BaSO₄ consisted of large foam cells on the top, the tiny foam cells in the middle, and the non-foamed region on the bottom (Figure 3 (a)). Due to foams transforming into solid-state foams slowly, it is easier to diffuse gas into the top of the sample, which leads to foam collapse and increased density on the bottom [14]. On the other hand, samples with increasing BaSO₄ contents showed increasing large voids since foams transformed to solid-state foams rapidly. Gas phases decrease tensile strength, elongation at break, and tear strength [3].

3.2 Mechanical properties

As shown in Figure 5, the tensile strength, elongation at break, and tear strength decreased with increasing BaSO₄ content. These were due to more BaSO₄ content being added to the composites, leading to an agglomeration of BaSO₄ particles and worse particle dispersion[16]. On the other hand, hardness (Shore A) increased with increasing BaSO₄ contents. The increases in hardness could have been due to the high rigidity of the BaSO₄ particles. This result was consistent with Lim-aroon et al. (2019) [16], which prepared X-ray and gamma-ray shielding from NR/Bi₂O₃ composites.

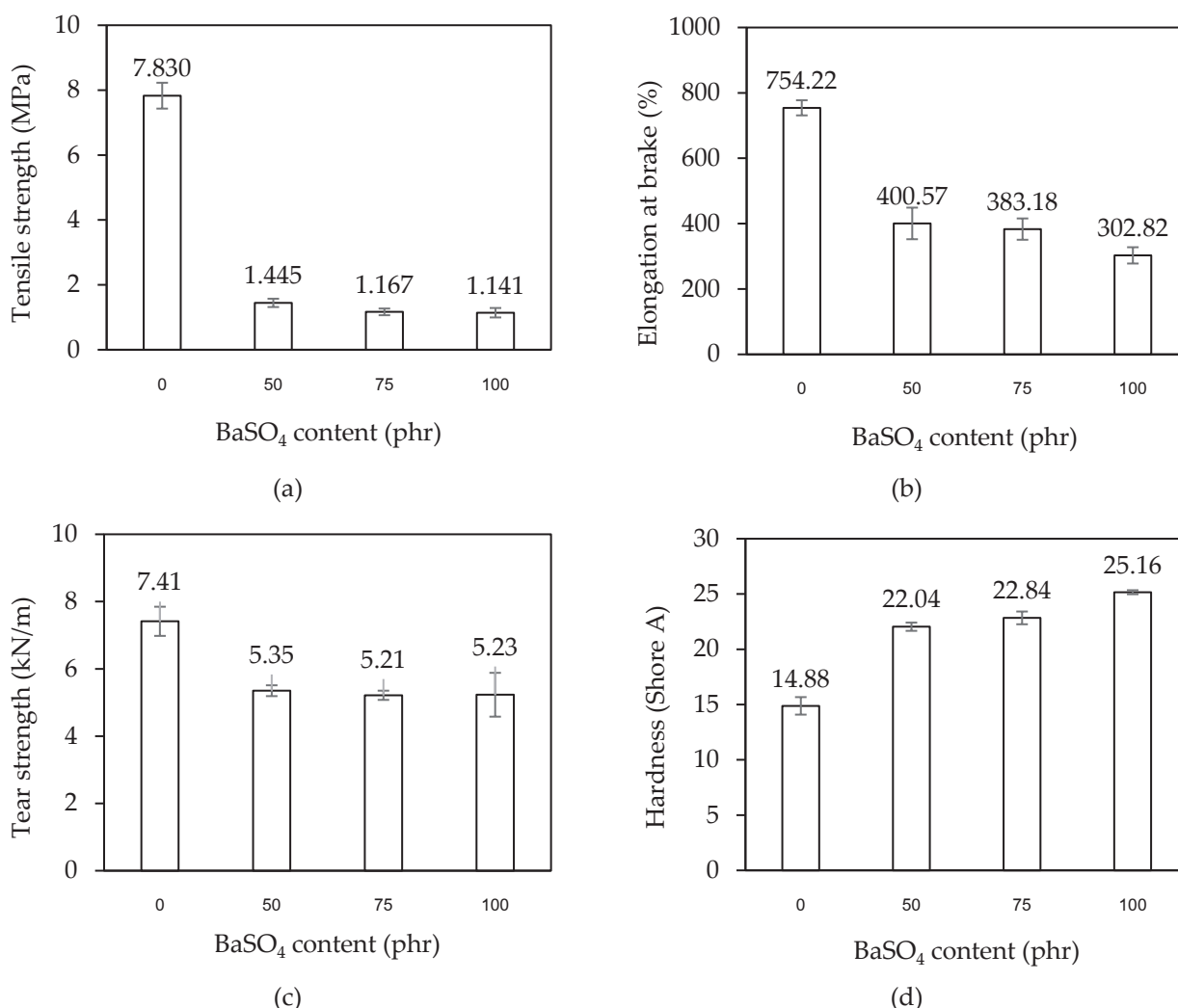


Figure 5. The mechanical properties of NR/BaSO₄ composites with varying contents of BaSO₄; (a) tensile strength, (b) elongation at break, (c) tear strength, and (d) hardness.

3.3 X-Ray shielding properties

Figure 6 shows the X-ray attenuation as a function of thickness for NR/BaSO₄ composites. It can be seen that the X-ray attenuation is higher compared to a pure NR. In addition, the X-ray attenuation increases with BaSO₄ content and thickness of samples due to the more remarkable ability of X-ray interaction and the impact of the photoelectric effect or Compton scattering with bismuth atoms in the material to attenuate and reduce X-ray energies and intensities [2]. For 2 cm thickness, the X-ray attenuation is highest at about 94% in 100 BaSO₄. From this result, it was possible to develop for actual use.

X-ray shielding properties of NR/BaSO₄ composites are shown in Table 2. The results showed that μ_l and μ_m increased, while HVL decreased with increasing BaSO₄ content. Causes may include the high Z number material (BaSO₄) [11], periodically distributed in the Na/BaSO₄ composites, which can act as an X-ray moderator resulting in μ_l improved HVL. At 100 phr of BaSO₄ content, the highest μ_l , μ_m , and the lowest HVL were 1.342 cm⁻¹, 1.923 cm²/g, and 0.517 cm, respectively. These results confirm that increasing the contents of BaSO₄ could improve the radiation shielding properties of NR/BaSO₄ composites.

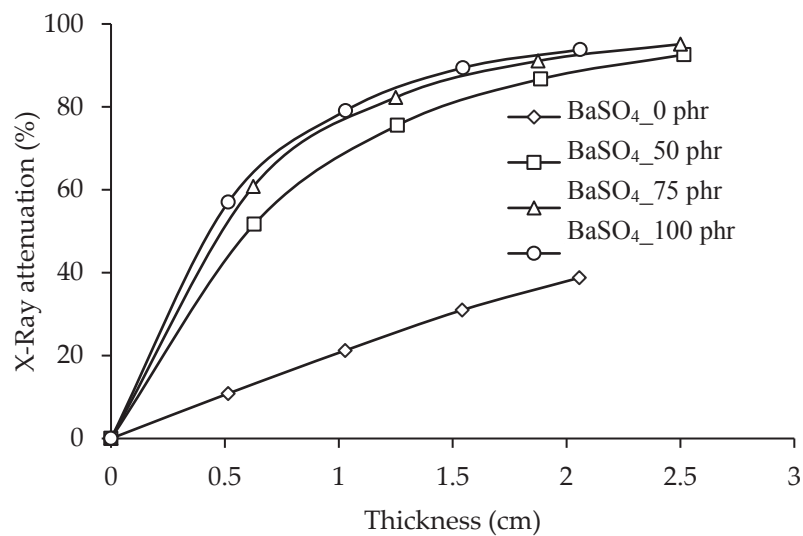


Figure 6. X-ray attenuation (%) of NR/BaSO₄ composites with varying contents of BaSO₄.

Table 2. The linear attenuation coefficient, mass attenuation coefficient and half-value layer of NR/BaSO₄ composites.

BaSO ₄ content (phr)	Linear attenuation coefficient; μ_l (cm ⁻¹)	Mass attenuation coefficient; μ_m (cm ² /g)	Half value layer; HVL (cm)
0	0.241	0.406	2.880
50	1.092	1.79	0.635
75	1.203	1.891	0.576
100	1.342	1.923	0.517

4. Conclusions

This work developed flexibility, lead-free, X-ray shielding materials based on NR/BaSO₄ composites with the contents varied BaSO₄ from 0 to 50, 75, and 100 phr. The results showed that BaSO₄ worked effectively as active fillers to shield X-rays. The NR/BaSO₄ composites had a higher linear attenuation coefficient (μ_l) and mass attenuation coefficient (μ_m) with increasing BaSO₄, while the HVL was decreased. Furthermore, it was found that the X-ray attenuation property improved with increasing BaSO₄ content and material thickness. The results also indicated that the increasing BaSO₄ increased the density and hardness of NR/BaSO₄ composites.

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